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Chiral Bicyclo[3.3.0]octa-2,5-dienes as Steering Ligands in Substrate-Dependent Rhodium-Catalyzed 1,4-Addition of Arylboronic Acids to Enones

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Dedicated to Prof. Peter Hofmann on the occasion of his 60th birthday.

Abstract: The synthesis of disubstituted chiral diene ligands (3aR,6aR)- and (3aS,6aS)-10 with a pentalene backbone from the corresponding bicyclo[3.3.0]octa-1,4-diones 7 is described. The latter were accessible by enzymatic resolution of the racemic diol *rac-5* and subsequent Swern oxidation. The efficiency of the ligands 10 in the rhodium-catalyzed 1,4-addition of arylboronic acids 12 to cyclic and acyclic enones 11 and 15 could be demonstrated. In the case of cyclic enones 11 the enantiomeric diphenyldienes

(3a*R*,6a*R*)- and (3a*S*,6a*S*)-**10a** were more selective than the corresponding dibenzyldiene ligands **10b**. When acyclic enones **15** were employed this result, however, reversed. Ligands **10a** were nearly inactive whereas dibenzyldienes **10b** afforded the addition products **16** with enantioselectivities up to 91%.

Keywords: asymmetric catalysis; chirality; diene ligands; enones; rhodium; synthesis

Introduction

In asymmetric catalysis alkenes such as ethylene, propene, 1,4-cyclooctadiene or norbornadiene and derivatives thereof were mainly considered as weak ligands, which do not allow stereochemical control in contrast to P-, N-, O- and S-containing ligands as well as N-heterocyclic carbenes. This situation changed with the recent seminal work of Hayashi, [1] Carreira, [2] and Grützmacher.[3] Their work confirmed the successful application of chiral diene ligands 1, 2 based bicycloalkanes, or tropylidenephosphane (Scheme 1), resulting in high selectivities, [4] for example, in the Ir-catalyzed hydrogenation of imines^[3d] and allylic substitution, [2c] in the Rh-catalyzed 1,4-addition of phenylboronic acid to enones[1h,l,2d] and fumarates^[1k] or in the Rh-catalyzed aryl transfer to sulfonylimines.[1j]

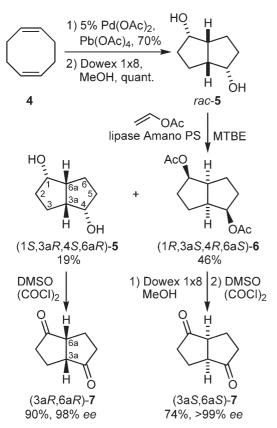
In some cases, however, the synthetic access to the ligands is rather tedious. [1h,k,l] Upon looking for alternative scaffolds which might be easily converted into chiral diene ligands in both enantiomeric forms, we turned our attention to C_2 -symmetrical bicyclo-[3.3.0]octa-1,4-dione 7 (Scheme 2), whose (3aR,6aR)-enantiomer has been used as a building block in our

total synthesis of the tetramic acid lactam cylindramide. We anticipated that the *convex* roof shape of dione $\mathbf{7}^{[6]}$ should lead to a diene ligand which bears structural similarities with the known norbornadiene ligands $\mathbf{1}^{[1f,5]}$ Further motivation came from a report by Trauner that *convex* phenanthrene ligands resulted in a chiral Pd(0) alkene complex with remarkable sta-

R
1 (R = Ph, Bn)
OMe
$$R^2$$
 R^2
 R

Scheme 1. Some chiral diene **1,2** and phosphane olefin ligands **3** reported by Hayashi, Carreira, and Grützmachar [1-3]

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Scheme 2. Preparation of chiral bicyclo[3.3.0]octadiones 7 via enzymatic resolution.

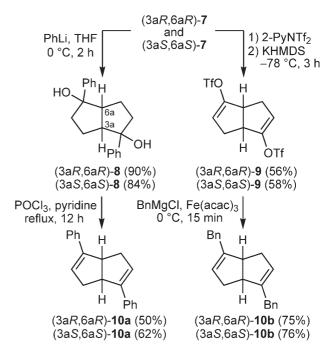
bility.^[7] Here we report the synthesis of bicyclo-[3.3.0] octadiene ligands and their application in the Rh-catalyzed 1,4-addition of arylboronic acids to enones.[8] Particular emphasis was spent on the influence of the diene substituents on the substrate specificity. The results are described below.

Results and Discussion

Ligand Synthesis

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As depicted in Scheme 2, the bicyclo[3.3.0]octa-1,4diones (3aR,6aR)- and (3aS,6aS)-7 were available starting from cycloocta-1,5-diene 4 by a literature procedure. [5] After transannular Pd-catalyzed ring closure^[9] and removal of the acetyl groups,^[10] the racemic diol rac-5 was obtained. Diol rac-5 was then submitted to enzymatic resolution with lipase Amano PS and vinyl acetate in methyl tert-butyl ether, [5,10,11] as a key step, yielding the diol (1S,3aR,4S,6aR)-5 and the diacetate (1R,3aS,4R,6aS)-6 in 19% and 46%, respectively. The diacetate 6 was separated by chromatography on SiO₂ and deacetylated to the corresponding alcohol (1R,3aS,4R,6aS)-5 by treatment with Dowex $1 \times$ 8 in MeOH for 24 h. Swern oxidation^[5,12] of both en-



Scheme 3. Preparation of chiral 3,6-disubstituted bicyclo-[3.3.0]octa-2,5-dienes **10a,b**.

antiomers 5 with DMSO and (COCl), afforded the diones (3aR,6aR)- and (3aS,6aS)-7 with 98% ee and >99% ee, respectively (Scheme 2).

As outlined in Scheme 3, the disubstituted bicyclo-[3.3.0]octa-2,5-dienes 10 were prepared analogously to literature procedures. The synthesis of diphenyldiene ligands (3aR,6aR)- and (3aS,6aS)-10a utilized the reaction of both enantiomers 7 with PhLi to give intermediates 8 in 90% and 84% yield, respectively. After dehydration with POCl₃ in pyridine at 80°C, [1f] ligands (3aR,6aR)- and (3aS,6aS)-10a were isolated in 50% and 62% yield. Deprotonation of diones (3aR,6aR)- and (3aS,6aS)-7 with KHMDS^[13] and subsequent treatment with N-(2-pyridyl)triflimide^[14] (2-PvNTf₂) in THF at -78°C gave the bistriflates (3aR,6aR)- and (3aS,6aS)-9 in 56–58% yield. The following Negishi coupling^[15] was performed with Fe(acac)₃ as the catalyst under the cross-coupling conditions recently described by Hayashi.[13] In this chiral 3,6-dibenzylated manner. the bicyclo-[3.3.0] octadiene ligands (3aR,6aR)- and (3aS,6aS)-10b were accessible in 75% and 76% yield, respectively. This synthesis route allows a variation of the substituent, yielding, for example, also ligand (3aS,6aS)-10a in 62% (see Experimental Section).

Crystallization of (3aS,6aS)-10a from MeOH/ CH₂Cl₂ gave single crystals which were suitable for Xray crystal structure analysis.^[16] Figure 1 clearly shows the roof-shape of the bicyclooctadiene with an angle of 115° for the convex bicyclic system. The double bonds between C-1/C-2 and C-5/C-6 are not parallel to each other but twisted by 28°, thus being similar to

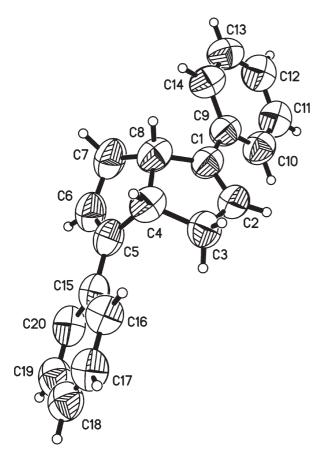


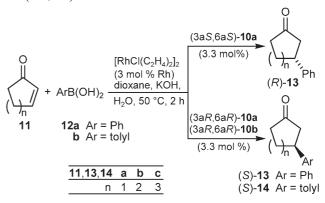
Figure 1. ORTEP plot of ligand 3,6-diphenylbicyclo-[3.3.0]octa-2,5-diene (3a*S*,6a*S*)-**10a**.

Hayashi's 2,6-di(4-methylphenyl)bicyclo[3.3.1]nona-2,6-diene rhodium complex where a twist angle of 23° has been found.^[1f]

Asymmetric Rh-Catalyzed 1,4-Addition

First the catalytic properties of the diene ligands 10 were investigated in the Rh-catalyzed 1,4-addition of arylboronic acids 12a, b to the cyclic enones 11a-c (Table 1). Treatment of cyclopentenone 11a with phenylboronic acid 12a in the presence of 3 mol% Rh and 3.3 mol% diphenyldiene ligand (3aS,6aS)-10a and KOH (0.5 equivs.) in aqueous dioxane yielded (R)-3-phenylcyclopentanone (R)-13a in 74% yield and 95% ee (entry 1). The enantiomeric ligand (3aR,6aR)-10a gave the corresponding (S)-configurated product in 80% yield and 93% ee (entry 2). The enantioselectivities decreased with both (3aS,6aS)and (3aR,6aR)-10a when cyclohexenone 11b and cycloheptenone 11c were employed, while, with exception of product 13c, good chemical yields were obtained (entries 4, 5, 7, 8). The addition of tolylboronic acid 12b in the presence of ligand (3aR,6aR)-10a proceeded with similar yields and enantioselectivities

Table 1. Rhodium-catalyzed addition of arylboronic acids **12a,b** to cyclic enones **11a–c** using ligands (3a*R*,6a*R*)-**10a,b** and (3a*S*,6a*S*)-**10a**.



Entry	Enone	Ligand	Product	Yield [%] ^[a]	% ee ^[b]	Conf. ^[c]
1	11a	(3aS,6aS)- 10a	13a	74	95	R
2	11a	(3a <i>R</i> ,6a <i>R</i>)- 10a	13a	80	93	S
3	11a	(3a <i>R</i> ,6a <i>R</i>)- 10b	13a	73	75	S
4	11b	(3a <i>S</i> ,6a <i>S</i>)- 10a	13b	95	88	R
5	11b	(3a <i>R</i> ,6a <i>R</i>)- 10a	13b	93	84	S
6	11b	(3a <i>R</i> ,6a <i>R</i>)- 10b	13b	78	73	S
7	11c	(3a <i>S</i> ,6a <i>S</i>)- 10a	13c	47	83	R
8	11c	(3a <i>R</i> ,6a <i>R</i>)- 10a	13c	51	76	S
9	11a	(3a <i>R</i> ,6a <i>R</i>)- 10a		85	93	$S^{[d]}$
10	11b	(3a <i>R</i> ,6a <i>R</i>)- 10a		81	84	$S^{[d]}$
11	11c	(3a <i>R</i> ,6a <i>R</i>)- 10a	14c	71	76	$S^{[\mathrm{d}]}$

[a] Yields are referred to isolated products.

(entries 9–11). A considerable drop of the enantioselectivity was observed when dibenzyldiene ligand (3aR,6aR)-10b was used under identical reaction conditions (entries 3, 6).

The most remarkable observation, however, was made during 1,4-addition of boronic acids **12** to acyclic enones **15** (Table 2).

As can be seen in Table 2, diphenyl- and dibenzyl-substituted ligands **10a** and **10b** showed a strikingly different activity. Whereas diphenyldiene ligand (3aS,6aS)-**10a** and its congener (3aR,6aR)-**10a** yielded

[[]b] Enantiomeric excess was determined by GC using chiral stationary phases.

[[]c] Assignment of the configuration by comparison of optical rotation values with literature data. [17,18]

 $^{^{[}d]}$ In analogy to the assignment of products 13.

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Table 2. Rhodium-catalyzed addition of arylboronic acids **12a,b** to acyclic enones **15a,b** using ligands (3aR,6aR)- and (3aS,6aS)-**10a,b**.

Enone	Ligand	Product	Yield [%] ^[a]	% ee ^[b]	$[\alpha]_{D}^{20}$ (c 1.0, CH ₂ Cl ₂)
15a	(3aS,6aS)- 10a	16a	<1 ^[c]	1	-
15a	(3a <i>R</i> ,6a <i>R</i>)- 10a	16a	<1 ^[c]	0	-
15a	(3a <i>S</i> ,6a <i>S</i>)- 10b	(S)- 16a ^[d]	71	91	+29.1
15a	(3a <i>R</i> ,6a <i>R</i>)- 10b	(R)- 16a ^[d]	59	89	-31.2
15b	(3a <i>S</i> ,6a <i>S</i>)- 10a	(+)- 16b ^[e]	<1 ^[c]	16	-
15b	(3a <i>R</i> ,6a <i>R</i>)- 10a	(-)- 16b ^[e]	1 ^[c]	10	-
15b	(3aS,6aS)- 10b	(+)- 16b	73	89	+63.4
15b	(3a <i>R</i> ,6a <i>R</i>)- 10b	(-)- 16b	45 ^[f]	88	-49.5
15a	(3a <i>R</i> ,6a <i>R</i>)- 10b	(R) - 17 $a^{[g]}$	68	89	-30.0
15b	(3a <i>R</i> ,6a <i>R</i>)- 10b	(-)- 17b	58	86	-65.2

- [a] Yields are referred to isolated products.
- [b] Enantiomeric excess was determined by GC using chiral stationary phases.
- [c] Starting material **15a** and **15b** was recovered.
- [d] Assignment of the configuration by comparison of optical rotation values with literature data.^[19]
- [e] According to the direction of rotation.
- [f] 72% yield with 6 mol% Rh catalyst.
- [g] In analogy to the assignment of derivative **16a**.

only trace amounts of the corresponding 1,4-addition products **16a** and **16b** in racemic form, the corresponding dibenzyldienes (3aS,6aS)- and (3aR,6aR)-**10b** gave the target products **16a**, **b** in good yields and enantioselectivities of 88–91%. When tolylboronic acid **12b** was employed in the presence of dibenzyldiene ligand (3aR,6aR)-**10b** comparable results were obtained. [20]

Conclusions

We have demonstrated that chiral bicyclo-[3.3.0] octadiene ligands 10a and 10b which were easily accessible in optically pure form in a five (or six) step sequence from cycloocta-1,5-diene 4, could be used in the catalytic asymmetric 1,4-addition of arylboronic acids to enones. More importantly, the substitution pattern of the diene led to complementary activity and substrate specificity of the catalyst. Whereas diphenyldiene ligands 10a converted cyclic enones 11a-c with good yields and high enantioselectivities, it turned out to be almost completely inactive towards acyclic enones 15a, b. In contrast, dibenzyldiene ligands 10b gave decreased selectivities for cyclic enones 11 as compared to the diphenyldiene counterpart 10a. For acyclic enones 15, however, ligands 10b produced good yields and selectivities. Thus, it needs to be explored whether such complementary behaviour of ligands 10a, b is also observed for other catalytic reactions.

Experimental Section

General Remarks

Melting points (uncorrected) were determined on a Büchi SMP 20. Specific rotations were determined on a Perkin–Elmer 241 polarimeter. IR spectra: Bruker Vektor22. Mass spectra: Finnigan MAT 95, Varian MAT 711, and Bruker Daltonics mircOTOFq. NMR spectra: Bruker Avance 300 and Avance 500. The spectra were recorded with TMS as internal standard. TLC: Silica gel 60 F_{254} (Merck). Column chromatography: Fluka Kieselgel 60, grain size 40–63 μm . GC: Fisons HRGC MEGA 8560 and Carlo Erba Strumentazione HRGC 5300. Column and temperature programs are given below.

(3aR,6aR)- and (3aS,6aS)-Hexahydropentalene-1,4-diones (7)

Compounds 7 were prepared as described in ref.^[5] In order to obtain enantiomerically pure (3aS,6aS)-7, the diacetate (1R,3aS,4R,6aS)-6 obtained from enzymatic resolution with 96% *ee* was submitted to saponification followed by enzymatic acetylation and Swern oxidation.

(3aR,6aR)-1,4-Diphenyloctahydropentalene-1,4-diol (8)

CeCl₃·7 H₂O (2.1 g, 5.6 mmol) was heated under vacuum at $140\,^{\circ}$ C for 2 h and cooled. Then THF ($10\,\text{mL}$) was added and the suspension stirred for 1 h at room temperature. The suspension was cooled to $-78\,^{\circ}$ C, PhLi (3.1 mL, 5.6 mmol, 1.8 M in dibutyl ether) was added and the reaction mixture stirred for a further 1 h. After addition of a solution of (3aR,6aR)-7 (276 mg, 2 mmol) in THF (2 mL), the reaction mixture was warmed to $-40\,^{\circ}$ C over 6 h. The reaction was quenched with water (20 mL), the layers were separated,

and the aqueous layer was extracted with EtOAc (3× 50 mL). The combined organic layers were dried (MgSO₄), and the solvent was removed under vacuum. Chromatography of the residue on SiO₂ [PE/EtOAc, 4:1 \rightarrow 0:1, $R_{\rm f}$ (PE/ EtOAc 5:1)=0.6] afforded **8** as a white solid; yield 531 mg (90%); mp 192°C; $[\alpha]_D^{20}$: -53.7 (c 1.0, CH₂Cl₂). FT-IR (ATR): $\tilde{v} = 3326$ (s), 2966 (m), 2925 (m), 1734 (m), 1494 (m), 1459 (m), 1444 (m), 1372 (m), 1297 (m), 1249 (m), 1154 (m), 1071 (m), 1037 (m), 927 (m), 756 (s), 697 (s) cm⁻¹; ¹H NMR (300 MHz, CDCl₃): $\delta = 1.73-1.85$ (m, 2H, 3-H_a, 6- H_a), 2.08–2.39 (m, 6H, 2-H, 3- H_b , 5-H, 6- H_b), 2.94–3.07 (m, 2H, 3a-H, 6a-H), 3.64 (s, 2H, OH), 7.22-7.39 (m, 2H, p-H), 7.31–7.39 (m, 4H, o-H), 7.49–7.54 (m, 4H, m-H); ¹³C NMR (62.5 MHz, CDCl₃): $\delta = 20.9$ (C-3, C-6), 45.1 (C-2, C-5), 55.4 (C-3a, C-6a), 81.3 (C-1, C-4), 125.3 (o-C), 126.9 (p-C), 128.3 (*m*-C), 145.0 (*i*-C); GC-MS (EI): m/z (%)=295 (1) [M⁺+ H], 276 (15) $[M^+-H_2O]$, 258 (80) $[M^+-2H_2O]$, 230 (20), 156 (40), 143 (50), 129 (45), 115 (43), $[M^+-2H_2O-2C_6H_5+H]$, 90 (50), 77 (25) $[C_6H_5]$; ESI-MS: m/z = 317.1517 [M⁺], calcd. for $C_{20}H_{22}NaO_2$ 317.1512; anal. calcd. for C₂₀H₂₂O₂ (294.38): C 81.60, H 7.53; found: C 81.37, H 7.63.

(3aS,6aS)-1,4-Diphenyloctahydropentalene-1,4-diol (8): Yield: 84%; $[\alpha]_0^{20}$: +54.1 (c 1.0, CH₂Cl₂). The spectroscopic data are in accordance with those of (3aR,6aR)-8.

(3a*R*,6a*R*)-3,6-Diphenyl-1,3a,4,6a-tetrahydropentalene (10a)

To a solution of (3aR,6aR)-8 (264 mg, 0.90 mmol) in pyridine (1 mL) at room temperature was added POCl₃ (495 μL), and the reaction mixture refluxed for 12 h. After cooling to room temperature, the reaction was quenched with water (10 mL) and extracted with CH_2Cl_2 (3×20 mL). The combined extracts were washed with a 2M solution of NaOH, dried (MgSO₄) and concentrated. Chromatography on SiO_2 (PE/Et₂O, 500:1, R_f =0.1) afforded **10a** as a white solid; yield: 132 mg (50%); $[\alpha]_D^{20}$: -406 (c 1.0, CH₂Cl₂). FT-IR (ATR): $\tilde{v} = 2904$ (m), 2837 (m), 1494 (m), 1444 (m), 747 (s), 694 (s) cm⁻¹; 1 H NMR (500 MHz, CDCl₃): $\delta = 2.42$ (ddd, $J=17.8, 5.9, 3.0 \text{ Hz}, 2 \text{ H}, 3 \text{-H}_a, 6 \text{-H}_a), 2.88-2.97 \text{ (m, 2 H, 3-1)}$ H_b, 6-H_b), 4.03–4.10 (m, 2H, 3a-H, 6a-H), 6.00–6.04 (m, 2H, 2-H, 5-H), 7.18-7.26 (m, 2H, p-H), 7.27-7.37 (m, 4H, o-H), 7.41–7.47 (m, 4H, m-H); 13 C NMR (125 MHz, CDCl₃): $\delta =$ 38.6 (C-3, C-6), 48.3 (C-3a, C-6a), 124.4 (C-2, C-5), 126.3 (C-m), 126.9 (C-p), 128.4 (C-o), 135.9 (C-i), 145.1 (C-1, C-4); MS (CI, CH₄): m/z (%) = 258 (100) [M⁺], 243 (19), 230 (17), 215 (11), 202 (7), 178 (11), 167 (28), 154 (31), 141 (15), 128 (12), 117 (88), 102 (8), 91 (28), 77 (9) $[C_6H_5]$; HR-MS (CI, CH₄): m/z = 258.1409 [M⁺], calcd. for C₂₀H₁₈: 258.1409.

(3aS,6aS)-3,6-Diphenyl-1,3a,4,6a-tetrahydropentalene (10a): Yield: 62%; $[\alpha]_D^{20}$: +396 (c 1.0, Et₂O). The spectroscopic data are in accordance with those of (3aR,6aR)-10a.

(3aR,6aR)-4-{[(Trifluoromethyl)sulfonyl]oxy}-3,3a,6,6a-tetrahydropentalen-1-yl Trifluoromethane-sulfonate (9)

A solution of KHMDS (1.32 g, 6.64 mmol) in THF (12 mL) was slowly added to a solution of (3aR,6aR)-7 (400 mg, 3.00 mmol) and N-(2-pyridyl)-bis(trifluoromethanesulfonimide) (2-pyNTf₂) (2.49 g, 6.96 mmol) in THF (12 mL) at

-78°C, and the reaction mixture stirred for 3 h. Then a saturated solution of NaHCO₃ (10 mL) was added. The aqueous layer was extracted with pentane (2×50 mL), the combined organic layers were washed with a solution of 5% NaOH-H₂O, dried (MgSO₄) and concentrated. Chromatogaphy on SiO_2 (PE/EtOAc, 25:1, R_f =0.4) afforded triflate **9** as a colorless oil; yield: 675 mg (56%); $[\alpha]_D^{20}$: -45.2 (c 1.0, CH₂Cl₂). FT-IR (ATR): $\tilde{v} = 2936$ (m), 2871 (m), 1964 (m), 1659 (s), 1420 (vs), 1331 (s) cm⁻¹; ¹H NMR (500 MHz, CDCl₃): δ = 2.51 (ddd, J=17.4, 5.2, 2.6 Hz, 2H, 3-H_a, 6-H_a), 2.62–2.68 (m, 2H, 3-H_b, 6-H_b), 3.62–3.65 (m, 2H, 3a-H, 6a-H), 5.58– 5.66 (m, 2H, 2-H, 5-H); 13 C NMR (125 MHz, CDCl₃): δ = 30.3 (C-3, C-6), 44.1 (C-3a, C-6a), 115.3 (C-2, C-5), 118.6 (q, J=2.6, SO₂CF₃), 149.0 (C-1, C-4) ppm; GC-MS (CI, CH₄): m/z (%)=402 (1) [M⁺], 269 (9) [M⁺-CF₃O₂S], 162 (2), 135 (6), 119 (12), 99 (6), 91 (19), 77 (8), 69 (36) [CF₃], 64 (17) $[SO_2]$, 55 (100); HR-MS (CI, CH₄): m/z = 401.9647 [M⁺], calcd. for $C_{10}H_8F_6O_6S_2$: 401.9666.

(3aS,6aS)-4-{[(Trifluoromethyl)sulfonyl]oxy}-3,3a,6,6a-tetrahydropentalen-1-yl trifluoromethanesulfonate (9): Yield: 58%; $[\alpha]_D^{20}$: +47.1 (c 1.0, CH₂Cl₂). The spectroscopic data are in accordance with those of (3aR,6aR)-9.

(3aR,6aR)-2,5-Dibenzyl-1,3a,4,6a-tetrahydropentalene (10b)

BnMgCl (0.5 mL, 1 mmol, 2M in THF) was slowly added to a solution of (3aR,6aR)-9 (210 mg, 0.52 mmol) and Fe(acac)₃ (37 mg, 0.10 mmol, 5 mol %) in THF (5 mL) at 0 °C, and the reaction mixture stirred for 15 min. The reaction was then quenched at 0°C with a saturated solution of NH₄Cl. The aqueous layer was extracted with CH₂Cl₂ (3×20 mL), the combined extracts were dried (MgSO₄), and concentrated. Chromatography on SiO₂ (PE/Et₂O, 500:1, R_f =0.2) afforded **10b** as a colorless oil; yield: 111 mg (75%); $[\alpha]_D^{20}$: -49.5 (c 1.0, CH₂Cl₂). FT-IR (ATR): $\tilde{v} = 3025$ (m), 2902 (m), 2844 (m), 1601 (s), 1493 (m), 1452 (m), 1155 (s), 1072 (m), 1029 (m), 819 (m), 752 (m), 696 (s), 614 (m) cm⁻¹; ¹H NMR (500 MHz, CDCl₃): $\delta = 2.25-2.32$ (m, 2H, 3-H_a, 6-H_a), 2.40-2.49 (m, 2H, 3-H_b, 6-H_b), 3.17–3.22 (m, 2H, 3a-H, 6a-H), 3.26 (dd, J=15.4, 1.4 Hz, 2H, 7-H_a, 8-H_a), 3.47 (d, J=15.4 Hz, 2H, 7-H_b, 8-H_b), 5.13–5.16 (m, 2H, 2-H, 5-H), 7.08– 7.14 (m, 8H, o-H, m-H) 7.17–7.29 (m, 2H, p-H); ¹³C NMR (125 MHz, CDCl₃): $\delta = 35.9$ (C-3, C-6), 36.0 (C-7, C-8), 49.8 (C-3a, C-6a), 123.3 (C-2, C-5), 125.9 (C-p), 128.2 (C-m), 129.0 (C-o), 140.0 (C-i), 145.9 (C-1, C-4); GC-MS (CI, CH₄): m/z (%)=286 (40) [M⁺], 195 (74) [M⁺-C₇H₇], 178 (8), 167 (22), 153 (8), 129 (10), 117 (26), 91 (100) $[C_7H_7]$, 77 (8) $[C_6H_5]$, 65 (14), 51 (6); HR-MS (CI, CH₄): m/z = 286.1722 $[M^+]$, calcd. for $C_{22}H_{22}$: 286.1721.

(3aS,6aS)-2,5-Dibenzyl-1,3a,4,6a-tetrahydropentalene (10b): Yield: 76%; $[\alpha]_D^{20}$: +52.8 (c 1.0, CH₂Cl₂). The spectroscopic data are in accordance with those of (3aR,6aR)-10b.

(3aS,6aS)-2,5-Diphenyl-1,3a,4,6a-tetrahydropentalene (10a): According to ref., [1fij] from PhMgBr (0.5 mL, 1.49 mmol, 3M in Et₂O), PdCl₂(dppf) (1.88 mg, 2.50 µmol), and (3aS,6aS)-9 (100 mg, 0.25 mmol) at room temperature, reaction time 16 h; yield: 62%; $[\alpha]_D^{20}$: +379.5 (c 1.0, Et₂O).

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General Procedure for the Rhodium-Catalyzed Asymmetric 1,4-Addition of Arylboronic Acids 12 to Enones 11 and 15

A solution of [RhCl(C₂H₄)₂]₂ (3 mol % Rh) and the respective ligand 10 (3.3 mol%) in degassed dioxane (2 mL) was stirred at room temperature for 15 min. Then a degassed 1M solution of KOH (0.5 equivs.) was added, and the mixture stirred for a further 10 min. After addition of the respective enones 11 or 15 (1 equiv.) and arylboronic acids 12 (2 equivs.), the reaction mixture was stirred at 50 °C for 2 h. The reaction was then quenched with a saturated solution of NH₄Cl (5 mL), and the aqueous layer was extracted with Et₂O (3×20 mL). The combined organic layers were dried (MgSO₄) and the solvent removed under vacuum. The crude products 13, 14 or 16, 17 were purified by chromatography on SiO₂ with PE/EtOAc (10:1). GC: cyclic products 13a,b, **14a,b**: column Bondex un α (20 m×0.25 mm), 0.4 bar H₂; cyclic products 13c, 14c: column Bondex un β (20 m× 0.25 mm); acyclic products 16a,b, 17a,b: column Amidex C mono-2-undecamethylenepermethyl-β-cyclodextrin, 1.2% *N*-(4-trimethylenoxybenzoyl)-L-valine bornylamide) $(20 \text{ m} \times 0.25 \text{ mm}), 0.35 \text{ bar H}_2.$

According to this protocol the following addition products were obtained. The spectroscopic data of 13, 14 and 16a, 17a are in accordance with those in the literature. [17,21]

(S)-3-Phenylcyclopentanone (13a): $R_f = 0.25$; $[\alpha]_D^{20}$: -101.7(c 1.0, CH₂Cl₂) [with (3aR,6aR)-**10a**], $[\alpha]_D^{20}$: -73.5 (c 1.0, CH_2Cl_2 [with (3aR,6aR)-10b]; GC: $10^{\circ}C$ min⁻¹ gradient from 40 °C to 120 °C, 3 min at 120 °C, then 2 °C min-1 gradient to 200 °C, $t_{R1} = 17.04$ min, $t_{R2} = 17.39$ min (major enantio-

(R)-3-Phenylcyclopentanone (13a): $[\alpha]_{D}^{20}$: +96.4 (c 1.0, CH_2Cl_2) [with (3aS,6aS)-**10a**]; GC: $t_{R1} = 17.04$ min.

(S)-3-Phenylcyclohexanone (13b): $R_f = 0.3$; $[\alpha]_D^{20}$: -15.0 (c 1.0, CH_2Cl_2) [with (3aR,6aR)-**10a**]; $[\alpha]_D^{20}$: -14.1 (c 1.0, CH_2Cl_2) [with (3aR,6aR)-**10b**]; GC: 10 °C min⁻¹ gradient from 40 °C to 120 °C, 3 min at 120 °C, then 2 °C min⁻¹ gradient to 150 °C, then 10 °C min⁻¹ gradient to 200 °C, t_{R1} 19.57 min, t_{R2} = 20.02 min (major enantiomer).

(R)-3-Phenylcyclohexanone (13b): $[\alpha]_D^{20}$: +8.4 (c 1.0, CH_2Cl_2) [with (3aS,6aS)-**10a**]; GC: $t_{R1} = 19.57$ min.

(S)-3-Phenylcycloheptanone (13c): $R_{\rm f} \! = \! 0.3; \, [\alpha]_{\rm D}^{20} \! : -43.0 \, (c$ 1.0, CH₂Cl₂) [with (3aR,6aR)-**10a**]; GC: 0.5°C min⁻¹ gradient from 80°C to 130°C, then 10°C min⁻¹ gradient to 200 °C, $t_{R1} = 83.88 \text{ min (major enantiomer)}, t_{R2} = 84.88 \text{ min.}$

(*R*)-3-Phenylcycloheptanone (13c): $[\alpha]_D^{20}$: +41.4 (*c* 1.0, CH_2Cl_2) [with (3aS,6aS)-**10a**]; GC: t_{R2} = 84.88 min.

(S)-3-(4-Methylphenyl)cyclopentanone (14a): $R_{\rm f} = 0.2$; $[\alpha]_D^{20}$: -89.0 (c 1.0, CH₂Cl₂) [with (3aR,6aR)-**10a**]; GC: 10 °C min⁻¹ gradient from 40 °C to 120 °C, 3 min at 120 °C, then 2 °C min⁻¹ gradient to 200 °C, $t_{R1} = 22.17$ min, $t_{R2} = 22.45$ min (major enantiomer).

(S)-3-(4-Methylphenyl)cyclohexanone (14b): $R_f = 0.23$; $[\alpha]_{\rm D}^{20}$: -19.8 (c 1.0, CH₂Cl₂) [with (3aR,6aR)-**10a**]; GC: 10 °C min⁻¹ gradient from 40 °C to 120 °C, 3 min at 120 °C, then 2°C min⁻¹ gradient to 200°C, t_{R1} = 24.13 min, t_{R2} = 24.62 min (major enantiomer).

(S)-3-(4-Methylphenyl)cycloheptanone (14c): $R_f = 0.28$; $[\alpha]_D^{20}$: -29.0 (c 1.0, CH₂Cl₂) [with (3aR,6aR)-**10a**]; GC: 0.5°C min⁻¹ gradient from 80°C to 130°C, then 10°C min⁻¹ to 200 °C, $t_{R1} = 85.67$ min (major enantiomer), $t_{R2} = 86.54$ min.

(R)-4-Phenylpentan-2-one (16a): $R_f = 0.3$; $[\alpha]_D^{20}$: -31.2 (c 1.0, CH₂Cl₂) [with (3aR,6aR)-**10b**]; GC: 1 min at 40 °C, then 2 °C min⁻¹ gradient to 200 °C, $t_{R1} = 27.15$ min (major enantiomer), $t_{R2} = 27.71$ min.

(S)-4-Phenylpentan-2-one (16a): $[\alpha]_D^{20}$: +29.1 (c 1.0, CH_2Cl_2) [with (3aS,6aS)-**10b**]; GC: $t_{R2} = 27.71$ min.

(-)-4-(2-Furyl)-4-phenylbutan-2-one (16b): $R_{\rm f} = 0.28$; $[\alpha]_{D}^{20}$: -49.5 (c 1.0, CH₂Cl₂) [with (3aR,6aR)-**10b**]. FT-IR (ATR): $\tilde{v} = 2361$ (m), 2342 (m), 1715 (s), 1586 (m), 1504 (m), 1494 (m), 1453 (m), 1413 (m), 1359 (s), 1158 (s), 1079 (m), 1009 (s), 936 (m), 922 (m), 808 (m), 731 (m), 698 (s), 577 (m), 535 (m), 509 (s) cm⁻¹; ¹H NMR (300 MHz, CDCl₃): $\delta = 2.09$ (s, 3H, 1-H), 3.00 (dd, J = 7.3, 16.7 Hz, 1H, 3-H_a), 3.23 (dd, J = 7.5, 16.7 Hz, 1 H, 3-H_b), 4.59 (dd, J = 7.5, 7.3 Hz, 1H, 4-H), 5.99 (d, J=3.3 Hz, 1H, 3'-H), 6.27 (dd, J=1.9, 3.1 Hz, 1H, 4'-H), 7.19-7.32 (m, 6H, aryl-H, 5'-H); ¹³C NMR (125 MHz, CDCl₃): $\delta = 30.4$ (C-1), 40.2 (C-4), 48.4 (C-3), 105.7 (C-4'), 110.2 (C-3'), 126.9 (C-p), 127.7 (C-m), 128.6 (C-o), 141.6 (C-5'), 141.7 (C-i), 156.5 (C-2'), 206.2 (C-2); MS (EI): m/z (%) = 215 (8) [M⁺+H], 214 (50) [M⁺], 171 (12) $[C_{12}H_{11}O]$, 157 (100) $[C_{11}H_{9}O]$, 128 (20), 115 (9), 103 (18), 77 (6) $[C_6H_6]$, 43 (17); ESI-MS: m/z = 237.0886 [M+ Na]+, calcd. for C₁₄H₁₄NaO₂: 237.0884. GC: 10 min at 100°C, then 2°C min⁻¹ gradient to 160°C, 10 min at 160°C, then $10\,^{\circ}\text{C}$ min⁻¹ gradient to $200\,^{\circ}\text{C}$, $t_{\text{R1}} = 29.91$ min, $t_{\text{R2}} =$ 30.35 min (major enantiomer).

(+)-4-(2-Furyl)-4-phenylbutan-2-one (16b): $[\alpha]_D^{20}$: +63.4 (c 1.0, CH_2Cl_2) [with (3aS,6aS)-10b]; GC: $t_{R1}=29.91$ min.

(R)-4-(4-Methylphenyl)pentan-2-one (17a): $R_f = 0.25$; $[\alpha]_D^{20}$: -30.0 (c 1.0, CH₂Cl₂) [with (3aR,6aR)-**10b**]; GC: 1 min at 40 °C, then 2 °C min⁻¹ gradient to 200 °C, $t_{R1} = 31.85$ min (major enantiomer), $t_{R2} = 32.17$ min.

(-)-4-(2-Furyl)-4-(4-methylphenyl)butan-2-one $R_f = 0.25$; $[\alpha]_D^{20}$: -65.2 (c 1.0, CH₂Cl₂) [with (3aR,6aR)-**10b**]. FT-IR (ATR): $\tilde{v} = 2361$ (m), 1715 (s), 1588 (m), 1513 (m), 1415 (m), 1358 (m), 1158 (s), 1112 (m), 1009 (m), 966 (s), 884 (m), 802 (s), 732 (s), 547 (m), 518 (m), 505 (m) cm⁻¹ ¹H NMR (300 MHz, CDCl₃): $\delta = 2.08$ (s, 3H, 1-H), 2.29 (s, 3H, CH₃), 2.98 (dd, J=7.4, 16.7 Hz, 1H, 3-H_a), 3.21 (dd, J=7.4, 16.7 Hz, 1 H, 3-H_b), 4.55 (dd, J=7.4, 7.4 Hz, 1 H, 4-H), 5.96-5.99 (m, 1H, 3'-H), 6.27 (dd, J=1.8, 3.1 Hz, 1H, 4'-H), 7.07–7.15 (m, 4H, aryl-H), 7.28 (dd, J=0.7, 1.8, 1H, 5'-H); ¹³C NMR (125 MHz, CDCl₃): $\delta = 20.9$ (CH₃), 30.3 (C-1), 48.4 (C-3), 49.8 (C-4), 105.5 (C-4'), 110.1 (C-3'), 127.6 (Cm), 129.2 (C-o), 136.3 (C-p), 138.6 (C-5'), 141.7 (C-i), 156.5 (C-2'), 206.2 (C-2); MS (EI): m/z (%)=229 (8) [M⁺+H], 228 (40) [M⁺], 185 (6), 171 (100) [C₁₂H₁₁O], 141 (7), 128 (12), 117 (10), 115 (7), 77 (2) $[C_6H_6]$, 43 (5); ESI-MS: m/z =251.1043 [M+Na]⁺, calcd. for $C_{15}H_{16}NaO_2$: 251.1046. GC: 3 min at 100 °C, then 1.5 °C min⁻¹ gradient to 160 °C, then 10 °C min⁻¹ gradient to 200 °C, $t_{R1} = 34.99 \text{ min}, t_{R2} =$ 35.32 min (major enantiomer).

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References

- [1] a) N. Tokunaga, T. Hayashi, Adv. Synth. Catal. 2007, 349, 513-516; b) F.-X. Chen, A. Kina, T. Hayashi, Org. Lett. 2006, 8, 341-344; c) T. Hayashi, N. Tokunaga, K. Okamoto, R. Shintani, Chem. Lett. 2005, 34, 1480-1481; d) R. Shintani, K. Okamoto, T. Hayashi, Org. Lett. 2005, 7, 4757-4759; e) R. Shintani, T. Kimura, T. Hayashi, Chem. Commun. 2005, 3213-3214; f) Y. Otomaru, A. Kina, R. Shintani, T. Hayashi, Tetrahedron: Asymmetry 2005, 16, 1673-1679; g) Y. Otomaru, K. Okamoto, R. Shintani, T. Hayashi, J. Org. Chem. 2005, 70, 2503-2508; h) R. Shintani, W.-L. Duan, T. Nagano, A. Okada, T. Hayashi, Angew. Chem. 2005, 117, 4687-4690; Angew. Chem. Int. Ed. 2005, 44, 4611-4614; i) A. Kina, K. Ueyama, T. Hayashi, Org. Lett. 2005, 7, 5889-5892; j) N. Tokunaga, Y. Otomaru, K. Okamoto, K. Ueyama, R. Shintani, T. Hayashi, J. Am. Chem. Soc. **2004**, 126, 13584–13585; k) R. Shintani, K. Ueyama, I. Yamada, T. Hayashi, Org. Lett. 2004, 6, 3425-3427; 1) T. Hayashi, K. Ueyama, N. Tokunaga, K. Yoshida, J. Am. Chem. Soc. 2003, 125, 11508-11509.
- [2] a) J.-F. Paquin, C. R. J. Stephenson, C. Defieber, E. M. Carreira, Org. Lett. 2005, 7, 3821–3824; b) J.-F. Paquin, C. Defieber, C. R. J. Stephenson, E. M. Carreira, J. Am. Chem. Soc. 2005, 127, 10850–10851; c) C. Fischer, C. Defieber, T. Suzuki, E. M. Carreira, J. Am. Chem. Soc. 2004, 126, 1628–1629; d) C. Defieber, J.-F. Paquin, S. Serna, E. M. Carreira, Org. Lett. 2004, 6, 3873–3876.
- [3] a) P. Maire, F. Breher, H. Grützmacher, Angew. Chem. 2005, 117, 6483-6487; Angew. Chem. Int. Ed. 2005, 44, 6325-6329; b) P. Maire, T. Büttner, F. Breher, P. Le Floch, H. Grützmacher, Angew. Chem. 2005, 117, 6477-6481; Angew. Chem. Int. Ed. 2005, 44, 6318-6323; c) F. Läng, F. Breher, D. Stein, H. Grützmacher, Organometallics 2005, 24, 2997-3007; d) P. Maire, S. Deblon, F. Breher, J. Geier, C. Böhler, H. Rüegger, H. Schönberg, H. Grützmacher, Chem. Eur. J. 2004, 10, 4198-4205.
- [4] Review: a) L. Hintermann, Nachr. Chem. 2005, 53, 139–143; b) F. Glorius, Angew. Chem. 2004, 116, 3444–3446; Angew. Chem. Int. Ed. 2004, 43, 3364–3366.
- [5] a) N. Cramer, S. Laschat, A. Baro, H. Schwalbe, C. Richter, Angew. Chem. 2005, 117, 831–833; Angew. Chem. Int. Ed. 2005, 44, 820–822; b) N. Cramer, M. Buchweitz, S. Laschat, W. Frey, A. Baro, D. Mathieu, C. Richter, H. Schwalbe, Chem. Eur. J. 2006, 12, 2488–2503
- [6] a) Y.-W. Zhong, P. Tian, G.-Q. Lin, Tetrahedron: Asymmetry 2004, 15, 771–776; b) J. Pérard-Viret, A. Rassat, Tetrahedron: Asymmetry 1994, 5, 1–4; c) R. M. Moriarty, M. P. Duncan, R. K. Vaid, O. Prakash, Org. Synth., Coll. Vol. VIII, 1993, 43–46.
- [7] M. A. Grundl, J. J. Kennedy-Smith, D. Trauner, *Organometallics* 2005, 24, 2831–2833.

- [8] a) T. Hayashi, K. Yamasaki, Chem. Rev. 2003, 103, 2829–2844; b) K. Fagnou, M. Lautens, Chem. Rev. 2003, 103, 169–196.
- [9] a) M. A. Dyadchenko, G. A. Danilova, I. Shpanig, G. Shik, K. K. Pivnitskii, *Zh. Org. Khim.* 1990, 26, 2536–2542;
 b) P. M. Henry, M. Davies, G. Ferguson, S. Phillips, R. Restivo, *Chem. Commun.* 1974, 112–113.
- [10] a) K. Lemke, S. Ballschuh, A. Kunath, F. Theil, *Tetrahedron: Asymmetry* **1997**, 8, 2051–2055; b) R. M. Moriarty, M. P. Duncan, R. K. Vaid, O. M. Prakash, *Org. Synth.* **1990**, 68, 175–181.
- [11] Y.-W. Zhong, X.-S. Lei, G.-Q. Lin, *Tetrahedron: Asymmetry* **2002**, *13*, 2251–2255.
- [12] a) E. Debie, T. Kuppens, K. Vandyck, J. Van der Eycken, B. Van der Veken, W. Herrebout, P. Bultinck, *Tetrahedron: Asymmetry* 2006, 17, 3203–3218; b) F. Chaubet, M. Nguyen Van Duong, J. Courtieu, A. Gaudemer, A. Gref, A. L. Crumbliss, *Can. J. Chem.* 1991, 69, 1107–1116.
- [13] G. Berthon-Gelloz, T. Hayashi, J. Org. Chem. 2006, 71, 8957–8960.
- [14] D. L. Comins, A. Dehghani, Tetrahedron Lett. 1992, 33, 6299-6302.
- [15] E. Negishi, F. Liu, in: Metal-Catalyzed Cross-Coupling Reactions, (Eds.: F. Diederich, P. J. Stang), Wiley-VCH, Weinheim, 1998, pp 1–48.
- [16] Crystallographic data (excluding structure factors) for the structure reported in this paper has been deposited with the Cambridge Crystallographic Data Centre as supplementary publication no. CCDC-629398. Copies of the data can be obtained free of charge on application to CCDC, 12 Union Road, Cambridge CB2 1EZ, UK [fax.: (internat.) + 44 1223/336–033; e-mail: deposit@ccdc.cam.ac.uk].
- [17] Y. Takaya, M. Ogasawara, T. Hayashi, J. Am. Chem. Soc. 1998, 120, 5579-5580.
- [18] M. Pucheault, S. Darses, J.-P. Genêt, Eur. J. Org. Chem. 2002, 3552–3557.
- [19] F. López, S. R. Harutyunyan, A. J. Minnaard, B. L. Feringa, J. Am. Chem. Soc. 2004, 126, 12784–12785.
- [20] Z.-Q. Wang, C.-G. Feng, M.-H. Xu, G.-Q. Lin, J. Am. Chem. Soc. 2007, 129, 5336–5337.
- [21] a) S. E. Denmark, N. Amishiro, J. Org. Chem. 2003, 68, 6997-7003; b) C. S. Cho, S. Motofusa, K. Ohe, S. Uemura, J. Org. Chem. 1995, 60, 883-888; c) N. Kamigata, A. Satoh, M. Yoshida, Phosphorus, Sulfur Silicon Relat. Elem. 1989, 46, 121-129; d) Y. Takaya, M. Ogasawara, T. Hayashi, Tetrahedron Lett. 1999, 40, 6957-6961; e) A. Hajra, N. Yoshikai, E. Nakamura, Org. Lett. 2006, 8, 4153-4155; f) C. S. Cho, S. Motofusa, K. Ohe, S. Uemura, Bull. Chem. Soc. Jpn. 1996, 69, 2341-2348; g) M. Asaoka, S. Shima, N. Fujii, H. Takei, Tetrahedron 1988, 44, 4757-4766.